Structural elucidation by 1D/2D NMR, FTIR, and mass spectrometry analyses

Table 1. ¹H and ¹³C NMR analysis

¹ H NMR analysis		
Chemical shift (δ, ppm)	Assignment(s)	
1.2 (t)	$-CH^{1}_{2}$ of the cyclopropane ring	
1.6 (t)	$-CH^2_2$ of the cyclopropane ring	
2.5–2.9 (m)	>CH ³ - of the cyclopropane ring	
7.2–7.4 (m)	H^4/H^5 of benzene ring	
8.5–8.8 (br, s)	$-NH^6$ 2	
¹³ C NMR analysis		
12	$> C^3$ H– of the cyclopropane ring	
21	$-C^{1}H_{2}$ of the cyclopropane ring	
31	$-C^2$ H ₂ of the cyclopropane ring	
120–125	C^4 , C^5 , and C^6 of the benzene ring	
138	C^7 of the benzene ring	

br = broad, m = medium, s = singlet, and t = triplet

Peaks assignment:

 1 H NMR (CDCl₃, 0.05% v/v TMS, 300MHz, δ ppm): 1.2 (2H, t), 1.6 (2H, t), 2.5–2.9 (1H, m), 7.2–7.4 (4H, m), and 8.5–8.8 (br, s)

¹³C NMR (CDCl₃, 0.05% v/v TMS, 300MHz, δ ppm): 12 (>C³H–), 21 (-C¹H₂), 31 (-C²H₂), 120–125 (>C^{4/5/7}–), and 138 (C⁶)

Discussions of results:

Herein, the presence of two $-\text{CH}_2-$ protons (H¹ and H²) was confirmed from the ¹H NMR peaks at 1.2 and 1.6 ppm (Figure 1 and Table 1). In ¹H-¹H COSY spectrum, the cross peaks between 1.2 and 1.6 ppm confirm the direct interaction between these two protons, which eventually confirms the presence of $-\text{CH}_2-\text{CH}_2-$ in the molecule (Figure 2 and Table 1). Interestingly, peaks within 2.5–2.9 ppm interact with both H¹ and H², confirming the presence of the cyclopropane ring in the structure (Figure 2). Furthermore, the peaks within 7.2–7.4 ppm confirm the existence of the benzene ring with para substitutions. The presence of $-\text{NH}_2$ group was confirmed from the broad/ singlet peak within 8.5–8.8 ppm. In ¹³C NMR spectrum (with DEPT-135), the $-\textbf{C}^1\text{H}_2-/-\textbf{C}^2\text{H}_2-$ peaks were confirmed from the negative signals at 21/ 31 ppm (Figure 3 and Table 1). In this context, the attachment between the protons and the carbon atoms is confirmed from the ¹H-¹³C HETCOR analysis (Figure 4). Therefore, the benzene ring carbons are assigned to the ¹³C peaks within 120–125 ppm, while the peaks within 10–31 ppm are assigned to the aliphatic regions of the cyclopropane ring (Figure 4 and Table 1). Therefore, the presence of one cyclopropane ring, benzene ring, and $-\text{NH}_2$ group was confirmed in the structure of the molecule. Furthermore, the molecular ion peak at 133 corresponds to the structure of 4-cyclopropylaniline (Figure 5).

FTIR analysis

Table 2. FTIR analysis of 4-cyclopropylaniline

peaks (cm ⁻¹)	Assignment(s)
3467 and 3421 (br)	–NH ₂ group was confirmed from the broad peaks
2901 (w)	C–H str. of aromatic benzene ring
2619 and 2495 (w)	N–H <i>str</i> : of the amine salts
1950 (m)	Tertiary ammonium ion specific peak
1683 (s)	
1606 (m)	
1583 (s)	
1570 (m)	C-H <i>def.</i> of cyclopropane ring
1498 (s)	C–C str. of benzene ring
1464 (s)	>CH ₂ bending of cyclopropane ring
1443 (m)	In plane C-H bending and torsional C-C bending of
	the aromatic ring
1406 (m)	
1388 (m)	C-H rocking vib. of cyclopropane ring
1341 (s)	C-N str. of primary aromatic amine
1229 (m)	
1193 (m)	
1151 (m)	
1085 (m)	In plane C–H bending of benzene ring
1060 (m)	
1049 (m)	
1033 (m)	In plane C–H bending of benzene ring
1013 (m)	In plane C–H bending of cyclopropane ring
999 (m)	
936 (m)	C–C <i>str</i> : of cyclopropane ring
913 (m)	
897 (s)	
886 (s), 800 (s), and 754 (s)	C-H wagging of benzene ring
742 (s)	Out of the plane C-H bending of benzene ring
695 (s), 573 (m), 530 (m)	Ring in plane bending of benzene ring

Herein, the presence of –NH₂ group was confirmed from the N–H *str.* within 3467–3421 cm⁻¹, C–N *str.* of primary aromatic amine at 1341 cm⁻¹, tertiary ammonium ion specific peak at 1950 cm⁻¹, and N–H *str.* of the amine salts within 2619–2495 cm⁻¹ (Figure 6). The presence of cyclopropane ring was confirmed from the >CH₂ *bending* at 1464 cm⁻¹ and in plane >CH– *bending* at 1013 cm⁻¹ (Figure 6 and Table 2). The peaks at 2901, 1498, 1085, 1033, 886/800/754, and 742 cm⁻¹ were ascribed to the C–H *str.*, C–C *str.*, in plane C–H *bending*, C–H *wagging*, and out of the plane C–H *bending* of the benzene ring, respectively (Figure 6 and Table 2). Furthermore, the ring in plane *bending* of the benzene ring was assigned to 695/573/530 cm⁻¹ (Figure 6 and Table 2). In this context, the peak at 1443 cm⁻¹ was ascribed to the in plane C–H *bending* and torsional C–C *bending* of the aromatic ring (Table 2).

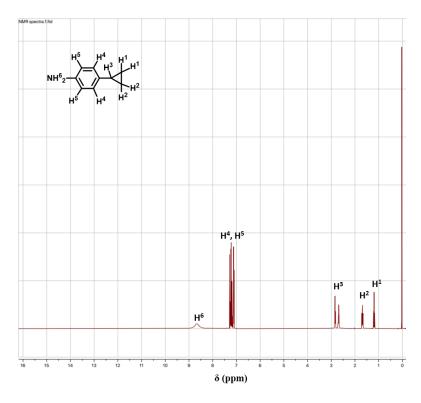


Figure 1. ¹H NMR spectrum of 4-cyclopropylaniline

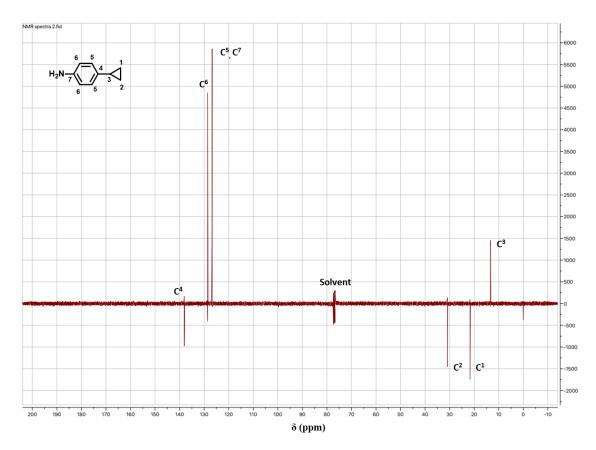


Figure 3. ¹³C NMR spectrum of 4-cyclopropylaniline

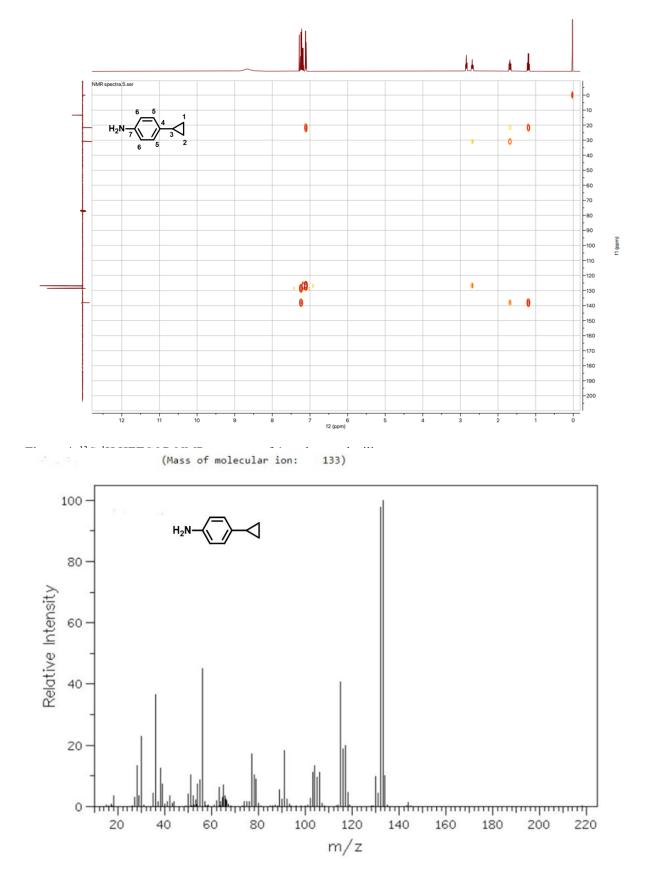


Figure 5. Mass spectrum of 4-cyclopropylaniline

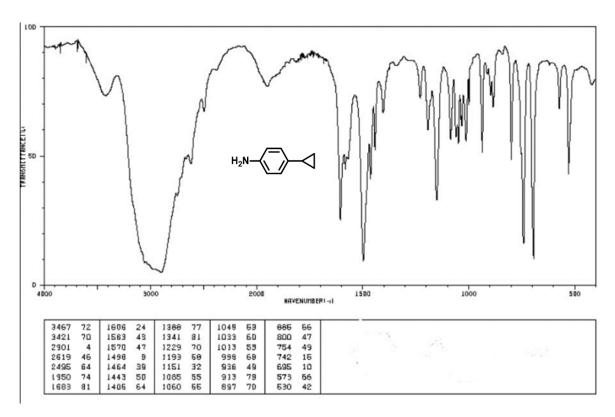


Figure 6. FTIR spectrum of 4-cyclopropylaniline